

NEW STANDARD ACADEMY

Test Type : Unit Test # 02

11-08-2025

Do not open this Test Booklet until you are asked to do so.

JEE(MAIN): 12th Undergoing/Pass Students

Read carefully the Instructions on the Back Cover of this Test Booklet.

Important Instructions :

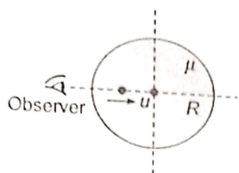
1. Immediately fill in the form number on this page of the Test Booklet with Blue/Black Ball Point Pen. Use of pencil is strictly prohibited.
2. The candidates should not write their Form Number anywhere else (except in the specified space) on the Test Booklet/Answer Sheet.
3. The Test Booklet consists of 45 questions
4. There are three parts in the question paper 1,2,3 consisting of Physics, Chemistry and Mathematics having 30 questions in each subject and each subject having Two sections. (i) Section-I contains 20 multiple choice questions with only one correct option. Marking scheme : +4 for correct answer, 0 if not attempted and -1 in all other cases. (ii) Section-II contains 10 Numerical Value Type questions. Attempt any 5 questions. First 5 attempted questions will be considered for marking. Marking scheme : +4 for correct answer, 0 if not attempted and -1 in all other cases.
5. Use Blue/Black Ball Point Pen only for writing particulars/markings responses on Side -1 and Side-2 of the Answer Sheet. Use of pencil is strictly prohibited.
6. No candidate is allowed to carry any textual material, printed or written, bits of papers, mobile phone any electronic device etc, except the Identity Card inside the examination hall/room.
7. Rough work is to be done on the space provided for this purpose in the Test Booklet only.
8. On completion of the test, the candidate must hand over the Answer Sheet to the invigilator on duty in the Room/ Hall. However, the candidate are allowed to take away this Test Booklet with them.

Name of the Candidate(In Capitals) _____

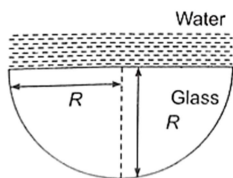
Date of Examination _____

Candidate's Signature: _____ Invigilator's Signature: _____

14. Consider a sphere of radius R made of glass of refractive index μ . A small object moves along the diameter with a constant velocity u . Find the velocity of the image as seen by an observer outside when the object passes through centre.



- (a) u (b) μu
 (c) u/μ (d) zero
15. A ray of light traveling in glass $\mu = \frac{3}{2}$ is incident on a horizontal glass air surface at the critical angle θ_c . If a thin layer of water ($\mu = \frac{4}{3}$) is now poured on the glass air surface, the angle at which the ray emerges into air at the water-air surface is



- (a) 60° (b) 45°
 (c) 90° (d) 180°

CHEMISTRY

16. IUPAC name of $[\text{Co}(\text{ONO})(\text{NH}_3)_5]\text{Cl}_2$ is
 (a) pentaamminenitrocobalt (III) chloride
 (b) pentaamminenitrito-o-cobalt (III) chloride
 (c) pentaamminenitrosocobalt (III) chloride
 (d) pentaammineoxo-nitrocobalt (III) chloride
17. How many isomers are possible in $[\text{Co}(\text{en})_2\text{Cl}_2]$
 (a) 2 (b) 4
 (c) 6 (d) 1
18. $[\text{Pt}(\text{NH}_3)_4]\text{Cl}_2$ is
 (a) Square planar (b) Tetrahedral
 (c) Pyramidal (d) Pentagonal
19. A tetrahedral complex ion is formed due to hybridization
 (a) sp^2 (b) sp^3
 (c) dsp^2 (d) d^2sp^3
20. The coordination and oxidation number of X in the compound $[\text{X}(\text{SO}_4)(\text{NH}_3)_5]\text{Cl}$ will be
 (a) 6 and 4 (b) 10 and 3
 (c) 2 and 6 (d) 6 and 3
21. The oxidation state of Fe in $\text{K}_4[\text{Fe}(\text{CN})_6]$ is
 (a) +2 (b) -2
 (c) +3 (d) +4

22. The number of moles of AgCl precipitate when excess of AgNO_3 is added to one mole of $[\text{Cr}(\text{NH}_3)_4\text{Cl}_2]\text{Cl}$ is
 (a) Zero (b) 1.0
 (c) 2.0 (d) 3.0
23. The number of unpaired electron in nickel carbonyl is
 (a) zero (b) one
 (c) four (d) five
24. Among the following ions which one has the highest paramagnetism?
 (a) $[\text{Cr}(\text{H}_2\text{O})_6]^{3+}$ (2) $[\text{Fe}(\text{H}_2\text{O})_6]^{2+}$
 (c) $[\text{Cu}(\text{H}_2\text{O})_6]^{2+}$ (4) $[\text{Zn}(\text{H}_2\text{O})_6]^{2+}$
25. Which of the following sequence is correct regarding field strength of ligands as per spectrochemical series?
 (a) $\text{SCN}^- < \text{F}^- < \text{CN}^- < \text{CO}$
 (b) $\text{F}^- < \text{SCN}^- < \text{CN}^- < \text{CO}$
 (c) $\text{CN}^- < \text{F}^- < \text{CO} < \text{SCN}^-$
 (d) $\text{SCN}^- < \text{CO} < \text{F}^- < \text{CN}^-$
26. $\text{K}_2[\text{OsCl}_5\text{N}]$ is named as:
 (a) Potassium pentachloroazooximate (VI)
 (b) Potassium pentachloronitridoosmate (VI)
 (c) Potassium pentachloroazidoosmate (VI)
 (d) Potassium pentachloronitroniumosmate (II)
27. Which of the following compounds shows optical isomerism?
 (a) $[\text{Cu}(\text{NH}_3)_4]^{2+}$ (b) $[\text{ZnCl}_4]^{2-}$
 (c) $[\text{Cr}(\text{C}_2\text{O}_4)_3]^{3-}$ (4) $[\text{Co}(\text{CN})_6]^{3-}$
28. Cuprammonium ion $[\text{Cu}(\text{NH}_3)_4]^{2+}$ is
 (a) tetrahedral (b) square planar
 (c) triangular bipyramid
 (d) octahedral
29. Which of the following compounds will exhibit highest magnetic moment?
 (a) $[\text{Ti}(\text{NH}_3)_6]^{3+}$ (b) $[\text{Cr}(\text{NH}_3)_6]^{3+}$
 (c) $[\text{Co}(\text{NH}_3)_6]^{3+}$ (d) $[\text{Zn}(\text{NH}_3)_6]^{2+}$
30. Consider the following two complex ions: $[\text{CoF}_6]^{3-}$ and $[\text{Co}(\text{C}_2\text{O}_4)_3]^{3-}$. Which of the following statement(s) is/are false?
 (A) Both are octahedral
 (B) $[\text{Co}(\text{C}_2\text{O}_4)_3]^{3-}$ is diamagnetic while $[\text{CoF}_6]^{3-}$ is paramagnetic
 (C) Both are outer orbital complexes
 (D) In both the complexes the central metal is in the same oxidation state
 (a) (B) and (C) (b) (B), (C) and (D)
 (c) (C) only (d) (C) and (D)

MATHS

31. The integral $\int \frac{(x^8 - x^2)dx}{(x^{12} + 3x^6 + 1)\tan^{-1}(x^3 + \frac{1}{x^3})}$ is equal
 (a) $\log_e \left(\left| \tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right| \right)^{1/3} + C$

- (b) $\log_e \left(\left| \tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right| \right)^{1/2} + C$
 (c) $\log_e \left(\left| \tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right| \right) + c$
 (d) $\log_e \left(\left| \tan^{-1} \left(x^3 + \frac{1}{x^3} \right) \right| \right)^3 + C$
32. $\int \frac{x^2-1}{(x^4+3x^2+1)\tan^{-1}(x+\frac{1}{x})} dx =$
 (a) $\tan^{-1} \left(x + \frac{1}{x} \right) + C$
 (b) $\left(x + \frac{1}{x} \right) \tan^{-1} \left(x + \frac{1}{x} \right) + c$
 (c) $\ln \left| \tan^{-1} \left(x + \frac{1}{x} \right) \right| + C$
 (d) $\frac{1}{2} \ln \left| \left(x + \frac{1}{x} \right) \right| + C$
33. $\int \frac{x+x^{\frac{2}{3}}+x^{\frac{1}{6}}}{x(1+x^{\frac{1}{3}})} dx$ equals
 (a) $\frac{3x^{\frac{2}{3}}}{4} + 6\tan^{-1} \left(x^{\frac{1}{6}} \right) + C$
 (b) $\frac{3x^{\frac{2}{3}}}{2} + 6\tan^{-1} \left(x^{\frac{1}{6}} \right) + C$
 (c) $\frac{3x^{\frac{2}{3}}}{10} + 6\tan^{-1} \left(x^{\frac{1}{6}} \right) + C$
 (d) $\frac{3x^{\frac{2}{3}}}{5} + 6\tan^{-1} \left(x^{\frac{1}{6}} \right) + C$
34. $\int \frac{1}{x} \ln \left(\frac{x}{e^x} \right) dx =$
 (a) $\frac{1}{2} e^x - \ln x + c$
 (b) $\frac{1}{2} \ln x - e^x + c$
 (c) $\frac{1}{2} (\ln x)^2 - x + c$
 (d) $\frac{e^x}{2x} + C$
35. $\int \frac{\{f(x) \cdot \varphi'(x) - f'(x) \cdot \varphi(x)\}}{f(x) \cdot \varphi(x)} \{ \log \phi(x) - \log f(x) \} dx$ is equal to
 (a) $\log \frac{\varphi(x)}{f(x)} + k$
 (b) $\frac{1}{2} \left\{ \log \frac{\varphi(x)}{f(x)} \right\}^2 + k$
 (c) $\frac{\varphi(x)}{f(x)} \log \frac{\varphi(x)}{f(x)} + k$
 (d) None of these
36. For any three positive real numbers a, b and $9(25a^2 + b^2) + 25(c^2 - 3ac) = 15b(3a + c)$.
 Then
 (a) a, b and c are in G.P.
 (b) b, c and a are in G.P.
 (c) b, c and a are in A.P.
 (d) a, b and c are in A.P.
37. If $a_1, a_2, a_3, \dots, a_n$ are in A.P. and $a_1 + a_4 + a_7 + \dots + a_{16} = 114$ then $a_1 + a_6 + a_{11} + a_{16}$ is equal to
 (a) 64 (b) 38
 (c) 76 (d) 98
38. If $3^{2\sin 2\alpha - 1}, 14$ and $3^{4 - 2\sin 2\alpha}$ are the first three terms of an A.P. for some α , then the sixth term of this A.P. is
 (a) 66 (b) 65
 (c) 81 (d) 78
39. For three positive integers p, q, r, $x^{pq^2} = y^{qr} = z^{p^2r}$ and $r = pq + 1$ such that $3, 3\log_y(x), 3\log_z(y), 7 \log_x(z)$ are in A.P. with common difference $1/2$ Then $r - p - q$ equal to
 (a) 6 (b) 2
 (c) 12 (d) -6
40. Let $A = \{1, 6, 11, 16, \dots\}$ and $B = \{9, 16, 23, 30, \dots\}$ be the sets consisting of the first 2025 terms of two arithmetic progressions. Then $n(A \cup B)$ is
 (a) 3761 (b) 3814
 (c) 4027 (d) 4003
41. Let a_1, a_2, \dots, a_n be a given A. P. whose common difference is an integer and $S_n = a_1 + a_2 + \dots + a_n$. If $a_1 = 1, a_n = 300$ and $15 \leq n \leq 50$ then the ordered pair (S_{n-4}, a_{n-4}) is equal to
 (a) (2480, 249) (b) (2490, 249)
 (c) (2490, 248) (d) (2480, 248)
42. The number of 4-digit numbers which are neither multiple of 7 nor multiple of 3 is _____
43. If $\int \frac{(1+\frac{1}{x})dx}{\sqrt{1+xe^x}} = \log \left| \frac{g(x)-1}{g(x)+1} \right| + C$, then $[g^2(1)]$ is equal to (where $[.]$ denotes the greatest integer function) _____
44. If $\log_3 2, \log_3 2^x - 5, \log_3 \left(2^x - \frac{7}{2} \right)$ are in an arithmetic progression, then the value of x is equal to _____
45. Let $a_1 = 8, a_2, a_3, \dots, a_n$ be an A.P. If the sum of its first four terms is 50 and the sum of its last four terms is 170, then the product of its middle two terms is _____